## General College Physics II (PHY 104)

Final Exam

Spring 2019

Instructions: Do 4 of the 6 questions and 3 of the 4 problems. Clearly indicate which questions and problems you want graded, or I will grade the first 4 questions and the first 3 problems.

## General College Physics II (PHY 104) Equation Sheet

$$F = k \frac{|qQ|}{r^2} \qquad PE = k \frac{qQ}{r}$$
$$F = qE \qquad PE = qV$$
$$E = k \frac{|Q|}{r^2} \qquad V = k \frac{Q}{r}$$
$$E = \frac{|\sigma|}{2\epsilon_0} = 2\pi k |\sigma| \qquad V = -2\pi k\sigma |x|$$

$$\Delta V = Ed \qquad \qquad Q = C\Delta V$$
$$C = \epsilon_0 \frac{A}{d} \qquad \qquad PE = \frac{1}{2}Q\Delta V = \frac{1}{2}C(\Delta V)^2 = \frac{1}{2}\frac{Q^2}{C}$$

$$V = IR \qquad \qquad P = IV = I^2R = \frac{V^2}{R}$$
$$R = \frac{\rho L}{A}$$

$$I_{\rm rms} = \frac{I_0}{\sqrt{2}} \qquad V_{\rm rms} = \frac{V_0}{\sqrt{2}} \\ R_{\rm eq} = R_1 + R_2 \qquad R_{\rm eq} = \frac{R_1 R_2}{R_1 + R_2}$$

$$F = \frac{\mu_0}{2\pi} \frac{I_1 I_2}{d} l$$

$$B = \frac{\mu_0 I}{2\pi r} \qquad B = \mu_0 N I / l$$

$$F = |q| v B \sin \theta \qquad F = I l B \sin \theta$$

$$\Phi_B = BA \cos \theta \qquad \mathcal{E} = -N \frac{\Delta \Phi_B}{\Delta t}$$

$$\Phi_B = BA \cos \omega t \qquad \mathcal{E} = N B \omega A \sin \omega t$$

$$n = \frac{c}{v} \qquad \qquad f = \frac{r}{2}$$

$$\frac{1}{d_o} + \frac{1}{d_i} = \frac{1}{f} \qquad \qquad m = \frac{h_i}{h_o} = -\frac{d_i}{d_o}$$

$$n_1 \sin \theta_1 = n_2 \sin \theta_2 \qquad \qquad \sin \theta_C = \frac{n_2}{n_1}$$

## **Conventions:**

- The focal length f is positive for concave mirrors and converging lenses, and negative for convex mirrors and diverging lenses.
- For a single-lens or single-mirror system, we choose  $d_o > 0$ .
- A real image has  $d_i > 0$ , while a virtual image has  $d_i < 0$ .
- An upright image has  $h_i > 0$ , while an inverted image has  $h_i < 0$ .

$$\lambda_n = \frac{\lambda}{n} \qquad \qquad \sin \theta = \frac{\lambda}{D}$$

$$\#\lambda s = \frac{d \sin \theta}{\lambda} \qquad \qquad \#\lambda s = \frac{2tn}{\lambda_0} + \left\{ \begin{array}{c} 1/2 \\ 0 \end{array} \right\} + \left\{ \begin{array}{c} 1/2 \\ 0 \end{array} \right\}$$

$$E = hf$$

$$p = \frac{L}{c} = \frac{n_f}{c} = \frac{\pi}{\lambda}$$

$$E_n = -(13.6 \text{ eV})\frac{Z^2}{n^2}$$

$$N = N_0 e^{-\lambda t} \qquad \qquad T_{\frac{1}{2}} = \frac{\ln 2}{\lambda}$$

$9.11 \times 10^{-31} \text{ kg}$
$1.6726 \times 10^{-27} \text{ kg}$
$1.6749 \times 10^{-27} \text{ kg}$
$1.6605 \times 10^{-27} \text{ kg} = 931.5 \text{ MeV}/c^2$
$6.02214 \times 10^{23} \text{ u/g}$
$1.602 \times 10^{-19} \text{ C}$
$k = 8.988 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2$
$\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2/\text{N} \cdot \text{m}^2$
$\mu_0 = 4\pi \times 10^{-7} \text{ T} \cdot \text{m/A}$
$G = 6.67 \times 10^{-11} \text{ N} \cdot \text{m}^2/\text{kg}^2$
$h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$
$c = 3.00 \times 10^8 \text{ m/s}$

 $1~{\rm eV} = 1.602 \times 10^{-19}~{\rm J}$ 

Material	Resistivity, $\rho$
Silver	$1.59 \times 10^{-8} \ \Omega \cdot m$
Copper	$1.68 \times 10^{-8} \ \Omega \cdot m$
Gold	$2.44 \times 10^{-8} \ \Omega \cdot m$
Tungsten	$5.6 \times 10^{-8} \ \Omega \cdot m$

Medium	Index of refraction, $n$
Vacuum	1.0000
Air (at STP)	1.0003
Water	1.33
Lucite	1.51
Crown glass	1.52

Earth's magnetic field in Annville, Pennsylvania  $B_x$  $20.2~\mu\mathrm{T}$ x is North  $B_y$  $-4.1 \ \mu T$ y is East  $B_z$ 47.4  $\mu T$ z is Down Horizontal Intensity  $20.6~\mu\mathrm{T}$  $51.7 \ \mu T$ Total Field Inclination (+D, -U)Declination (+E, -W) $66.5^{\circ}$  $-11.4^{\circ}$ 

From Giancoli 7th, Appendix B							
Mass							
		Number	Atomic	Half-life			
Element	Symbol	A	$Mass^{\dagger}$	(if radioactive)			
(Neutron)	n	1	1.008665	10.183 min			
Hydrogen	Н	1	1.007825				
proton	р	1	1.007276				
Deuterium	$^2_1\mathrm{H}$	2	2.014102				
Tritium	$^3_1\mathrm{H}$	3	3.016049	12.32 yr			
Helium	He	3	3.016029				
		4	4.002603				
Carbon	$\mathbf{C}$	11	11.011434	20.334 min			
		12	12.000000				
		13	13.003355				
		14	14.003242	5730 yr			
Sodium	Na	22	21.994437	2.6027 yr			
		23	22.989769				
		24	23.990963	14.997 hr			
Lead	Pb	206	205.974466				
		207	206.975897				
		208	207.976652				
		210	209.984189	22.20 yr			
		211	210.988737	36.1 min			
		212	211.991898	10.64 h			
		214	213.999806	26.8 min			
Polonium	Ро	210	209.982874	138.376 days			
		214	213.995202	164.3 $\mu {\rm s}$			
Astatine	At	218	218.008695	1.5 s			
	Element (Neutron) Hydrogen proton Deuterium Tritium Helium Carbon Sodium Lead Polonium Astatine	ElementSymbol(Neutron)nHydrogenHprotonpDeuterium1/1Tritium1/1HeliumHeCarbonCSodiumNaLeadPbPoloniumPoAstatineAt	Item chancel (uit, Ap)         Mass           Number         Number           Element         Symbol         A           (Neutron)         n         1           Hydrogen         H         1           proton         p         1           Deuterium <sup>2</sup> H         2           Tritium <sup>1</sup> H         3           Helium         He         3           Carbon         C         11           Sodium         Na         22           Lead         Pb         206           Polonium         Polonium         210           Astatine         At         218	Item Grancon (till, Appendix B)           Mass         Atomic           Element         Symbol         A         Mass <sup>†</sup> (Neutron)         n         1         1.008665           Hydrogen         H         1         1.007825           proton         p         1         1.007276           Deuterium <sup>2</sup> H         2         2.014102           Tritium <sup>3</sup> H         3         3.016049           Helium         He         3         3.016029           Helium         He         3         3.016029           Carbon         C         11         11.011434           Garbon         C         11         11.011434           Sodium         Na         22         21.994437           Sodium         Na         22         21.994437           Lead         Pb         206         205.974466           207         206.975897         208         207.976652           210         209.984189         211         210.988737           212         211.991898         214         213.999806           Polonium         Po         210         209.98274			

<sup>†</sup>Masses given are those for the neutral atom, including the Z electrons (except for the proton).

**Question 1** (4 points) Chapter 17 Question 5

Question 2 (4 points) Electric Circuits 4

**Question 3** (4 points) Magnetism 7

**Question 4** (4 points) Ch 23 MisConceptual Question 6

**Question 5** (4 points) Consider a convex mirror with radius 20 cm. An object is placed 5 cm in front of the mirror. Make a ray diagram that shows where an image would be formed. Is the image real or virtual? Is it inverted or upright?



**Question 6** (4 points) In Rutherford's planetary model of the atom, what keeps the electrons from flying off into space?

**Problem 1** (8 points) Chapter 16 Problem 11

**Problem 2** (8 points) Analyze the circuit below. Give the voltage across and the current through each circuit element listed in the box provided.



	Voltage	with high potential	Current	with current flowing
	across	on which side?	through	toward what direction?
		(left, right,		(left, right,
		top, bottom)		up,down)
$4 \Omega$				
$6 \Omega$				

**Problem 3** (8 points) Ch 24 Problem 52

**Problem 4** (8 points) Chapter 30 Problem 45